

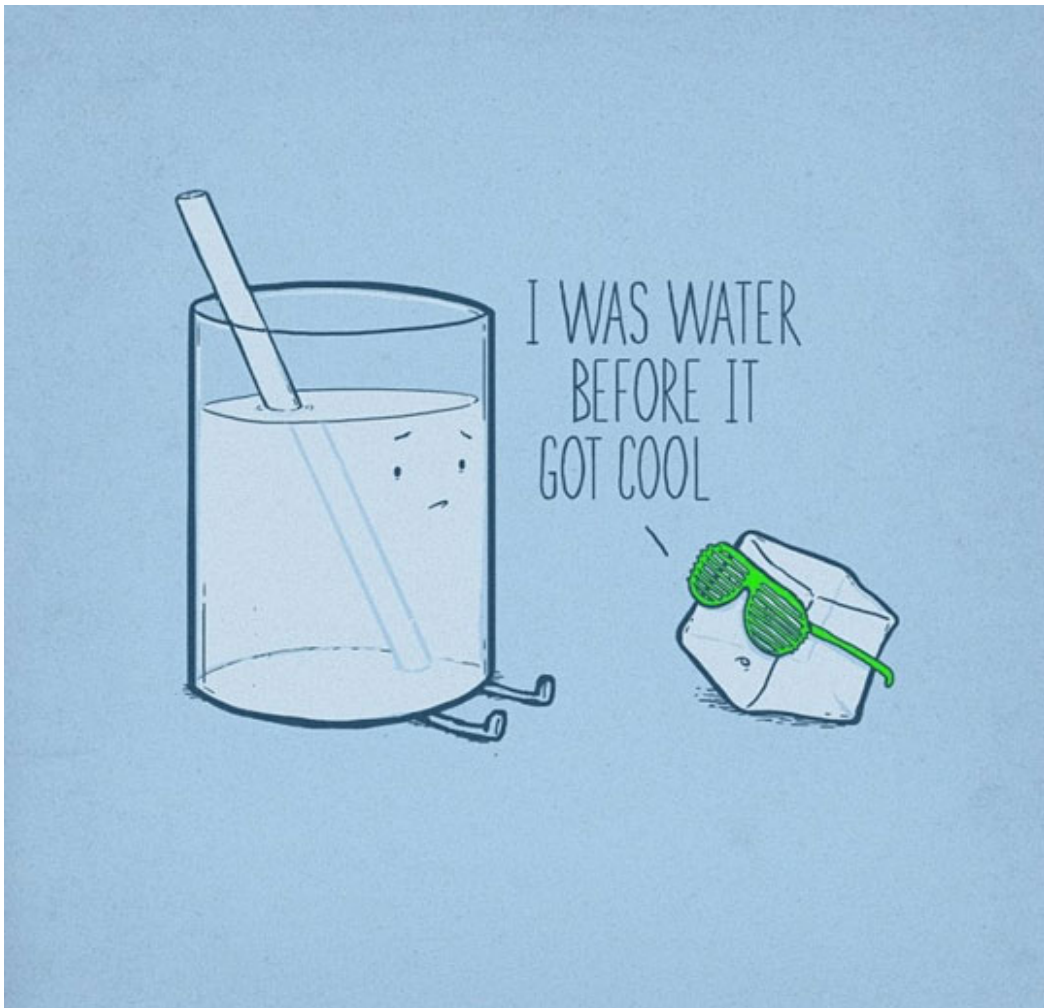
Name _____

GVC Science Dept.

Course: Science 10F

Intro to Lab Skills

Day 2: WaterWorld Lab



Introduction

The main purpose of the **WaterWorld Lab** is to continue working with basic lab techniques and skills. In this lab you will practice using scientific notation, measuring volume of a liquid, identifying common lab equipment, graphing data, and reviewing lab safety.

Materials

Materials for this lab include:

1. Rubber Stopper	2. Graduated Cylinders (3 sizes)	3. Electronic Balance	4. Beakers of liquid	5. Pipette	6. Erlenmeyer Flask
					7. Goggles

Procedure

Part A: In this section, you will review conversions from the last lab.

Complete each of the following, using the information you learned in the **Sandman Lab**.

1) 52 daL = _____ L

2) 17 mg = _____ cg

3) 364 g = _____ hg

4) 892 mL = _____ cL

To Change Units, Move the Decimal Point

10^3	10^2	10^1	10^0	10^{-1}	10^{-2}	10^{-3}
k	h	da	1	d	c	m
kilo	hecto	deka	UNIT	deci	centi	milli



2.3 km = _____ m

2.3 km = 2300 m

2.3 00.

Part B: Scientific notation explanation with examples will be shown, together with some questions to attempt.

Review of Scientific Notation

Scientific notation provides a place to hold the zeroes that come after a whole number or before a fraction. The number 100,000,000 for example, takes up a lot of room and takes time to write out, while 108 is much more efficient.

Though we think of zero as having no value, zeroes can make a number much bigger or smaller. Think about the difference between 10 dollars and 100 dollars. Even one zero can make a big difference in the value of the number. In the same way, 0.1 (one-tenth) of a government budget is much more than 0.01 (one-hundredth) of the budget.

The small number to the right of the 10 in scientific notation is called the exponent. Note that a negative exponent indicates that the number is a fraction (less than one).

The line below shows the equivalent values of decimal notation (the way we write numbers usually, like "1,000 dollars") and scientific notation (10³ dollars). For numbers smaller than one, the fraction is given as well.

	smaller		bigger			
Fraction	1/100	1/10				
Decimal notation	0.01	0.1	1	10	100	1,000
Scientific notation	10 ⁻²	10 ⁻¹	10 ⁰	10 ¹	10 ²	10 ³

Practicing Scientific Notation

Write out the decimal equivalent (regular form) of the following numbers that are in scientific notation.

Section A: Model: $10^1 = 10$

1) $10^2 =$ _____

4) $10^{-2} =$ _____

2) $10^4 =$ _____

5) $10^{-5} =$ _____

3) $10^7 =$ _____

6) $10^0 =$ _____

Section B: Model: $2 \times 10^2 = 200$

7) $3 \times 10^2 =$ _____

10) $6 \times 10^{-3} =$ _____

8) $7 \times 10^4 =$ _____

11) $900 \times 10^{-2} =$ _____

9) $2.4 \times 10^3 =$ _____

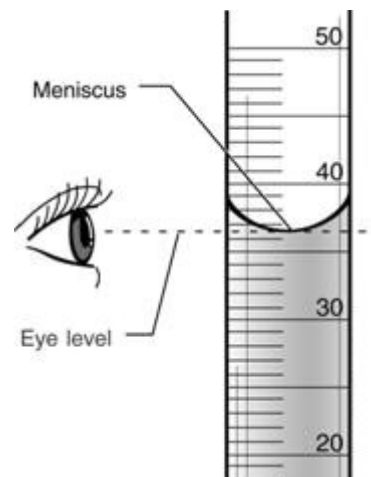
12) $4 \times 10^{-6} =$ _____

Name _____

Part C: In this section, you will accurately measure and record the volume of liquid substances.

Measuring the volume of a liquid using a graduated cylinder.

- Depending on the volume to be measured, you will need to use a small, medium or large graduated cylinder.
- Each graduated cylinder has lines on it signifying the units that are used on the cylinder.
- When measuring a water based solution, always measure from the lowest point or the bottom of the meniscus. Also make sure the graduated cylinder is on a level surface (table or counter) and get down to eye level with the liquid to get an accurate measurement.
- Measure out 2.4 mL in the small graduated cylinder.
- Measure out 25 mL in the large graduated cylinder.
- Measure out 0.150 L in a graduated cylinder



0.150 L = _____ mL

Part D: In this section, you are required to create a graph (on the graph paper provided). The graph must follow the instructions below, and be based on the data provided below.

Graphing Checklist:

- Title (appropriate with Data being graphed)
- Axis labels with quantities and units (x and y axis)
- The scale along each axis is chosen to spread out the plotted points on the page.
- Consistent intervals (distance between axis labels, eg. 2, 4, 6, 8)
- On a line graph, remember to connect the points, and on a curve ensure that your curve is drawn smoothly.

1.	1955-56	252.6 cm
2.	1915-16	216.7 cm
3.	1996-97	213.4 cm
4.	1909-10	212.8 cm
5.	1965-66	212.6 cm
6.	1919-20	211.8 cm
7.	1949-50	201.9 cm
8.	1893-94	194.0 cm
9.	1906-07	193.5 cm
10.	1935-36	184.3 cm

Part E: Continue working on the *Lab Safety Assessment*.

Name _____

